Total Synthesis of (±)-Mycoepoxydiene, a Novel Fungal Metabolite Having an Oxygen-Bridged Cyclooctadiene Skeleton

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ABSTRACT



The first total synthesis of (\pm)-mycoepoxydiene has been accomplished. A ring-closing olefin metathesis (RCM) approach was employed for the construction of the oxygen-bridged eight-membered bicyclic skeleton. The RCM product was converted to the target natural product featuring the oxidative rearrangement of a furfuryl alcohol introduced as the side chain and the stereoselective 1,2-reduction of a δ -keto- $\beta_i \gamma$ -unsaturated α -lactol intermediate.

Mycoepoxydiene (1) (Scheme 1) was isolated from the fermentation of a rare fungus designated as OS-F66617 in 1999.¹ Its gross structure and relative stereochemistry were determined by spectral and X-ray crystallographic analyses, but the absolute stereochemistry is still unknown.¹ This metabolic product was found to possess a structural novelty that consists of an oxygen-bridged cyclooctadiene core skeleton and a functionalized δ -lactone. Although oxygenbridged dibenzocyclooctadiene lignans such as kadsulignans are known,² compound $\mathbf{1}$ is, to the best of our knowledge, the only reported natural product containing a 9-oxabicyclo-[4.2.1]nona-2,4-diene skeleton without a fused biphenyl system. This unique structural feature prompted us to attempt the total synthesis of 1. Herein we report the first total synthesis of (\pm) -mycoepoxydiene featuring a ring-closing diene metathesis (RCM) approach to the construction of the core oxygen-bridged eight-membered carbocyclic skeleton.

(1) Cai, P.; McPhail, A. T.; Krainer, E.; Katz, B.; Pearce, C.; Boros, C.; Caceres, B.; Smith, D.; Houck, D. R. *Tetrahedron Lett.* **1999**, *40*, 1479–1482.

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Recently, RCM has received a great deal of attention in the synthesis of medium- and large-sized rings from acyclic dienes. A large number of applications to natural product synthesis have revealed the versatility of the RCM reaction.³ We envisaged the use of the RCM strategy as a means to



⁽²⁾ Liu, J.-S.; Li, L. Phytochemistry 1995, 38, 241-245.

access the core skeleton of $1.^4$ Our retrosynthetic analysis of 1 is shown in Scheme 1. The α,β -unsaturated δ -lactone moiety of 1 would be constructed at a later stage of total synthesis by employing the oxidative rearrangement of a furfuryl alcohol to a hydroxylated pyranone.⁵ The requisite furfuryl alcohol derivative 2 could be prepared using the coupling of an aldehyde derivative derived from bicyclic intermediate 3 with furan as a key step. The 9-oxabicyclo-[4.2.1]non-3-ene derivative 3 was expected to be obtained by the RCM reaction of diene 4, which in turn would be prepared from the known 5,⁶ the Diels—Alder adduct of furan and maleic anhydride.

The synthesis began with the preparation of 14 (P = TBDPS in 4), the substrate for the RCM reaction (Scheme 2). Monosilylation of *meso*-diol 6, prepared from 5 by a



^{*a*} Reagents and conditions: (a) NaH, TBDPSCl, THF, 92%; (b) NaH, CS₂, MeI, THF, 85%; (c) OsO₄, TMNO, aqueous acetone; (d) *n*-Bu₃SnH, AIBN, toluene, reflux, 88% for two steps; (e) Pb(OAc)₄, PhH; (f) NaBH₄, MeOH, 84% for two steps; (g) *n*-BuLi, TsCl, THF, 0 °C, 89%; (h) NaI, 2-butanone, reflux, 95%; (i) vinylMgBr, PhH, 80%.

known procedure,⁷ was conducted with sodium hydride and *tert*-butyldiphenylsilyl chloride (TBDPSCl). The resulting mono-*O*-TBDPS derivative **7** was converted to xanthate ester **8** using standard reaction conditions. The treatment of **8** with osmium tetroxide and trimethylamine-*N*-oxide (TMNO)⁸

underwent stereoselective dihydroxylation to give *exo*-diol **9**, which was subjected to the tri-*n*-butyltin hydride reduction of the xanthate ester to provide deoxygenated product **10** efficiently. The oxidative cleavage of the vicinal diol in **10** with lead tetraacetate followed by the sodium borohydride reduction of the resulting dialdehyde hydrate provided a tetrasubstituted tetrahydrofuran derivative **11**. The 1,4-carbinols in **11** were converted simultaneously to diiodo derivative **13** via the corresponding ditosylate **12**. Simultaneous substitution of the two iodo groups in **13**, each by a vinyl group, was achieved efficiently using an excess amount of vinylmagnesium bromide in benzene, affording 1,4-diallylated tetrahydrofuran **14**.⁹

With the designed diene **14** in hand, the RCM reaction was next examined (Scheme 3). The treatment of **14** with a



^{*a*} Reagents and conditions: (a) **15** (20 mol %), PhH, reflux, 83%; (b) Br₂, Et₂O, -78 °C, 88%; (c) *t*-BuOK, *t*-BuOH, 75 °C, 58%.

Grubbs catalyst 15^{10} (four additions, each of 5 mol % 15 over a period of 20 h) in refluxing benzene furnished the desired oxygen-bridged cyclooctene derivative 16 (P = TBDPS in 3) in an 83% yield.¹¹ In this case, a highly diluted solution (0.003 M) of 14 in benzene was essential to obtain a satisfactory result.¹² The addition of bromine to 16, followed by the exposure of the resulting vicinal dibromo adduct 17 as a diastereomeric mixture to potassium *t*-butoxide, provided a 9-oxabicyclo[4.2.1]nona-2,4-diene derivative 18.¹³

Having established the synthetic route to the oxygenbridged cyclooctadiene skeleton, we turned our attention to the construction of the δ -lactone moiety attached to the main scaffold (Scheme 4). The desilylation of **18** with tetra-*n*-

⁽³⁾ For recent reviews regarding the ring-closing metathesis, see: (a) Yet, L. Chem. Rev. 2000, 100, 2963–3007. (b) Fürstner, A. Angew. Chem., Int. Ed. 2000, 39, 3012–3043. (c) Grubbs, R. H.; Chang, S. Tetrahedron 1998, 54, 4413–4450. (d) Armstrong, S. K. J. Chem. Soc., Perkin Trans. I 1998, 371–388.

⁽⁴⁾ While this work was in progress, a metathesis strategy for preparing a 9-oxabicyclo[4.2.1]non-3-ene system was reported; see: Hanna, I.; Michaut, V. *Org. Lett.* **2000**, *2*, 1141–1143.

⁽⁵⁾ For a review regarding furan as a building block in synthesis, see: Maier, M. In *Organic Synthesis Highlights II*; Waldmann, H., Ed.; VCH: Weinheim, Germany, 1995; pp 231–242.

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⁽⁷⁾ Das, J.; Vu, T.; Harris, D. N.; Ogletree, M. L. J. Med. Chem. 1988, 31, 930–935.

⁽⁸⁾ Ray, R.; Matteson, D. S. Tetrahedron Lett. 1980, 21, 449-450.

⁽⁹⁾ When tetrahydrofuran was used as the solvent, the reaction proceeded less effectively. Neither vinylcopper reagents nor vinyllithium provided superior results.

^{(10) (}a) Schwab, P.; France, M. B.; Ziller, J. W.; Grubbs, R. H. Angew. Chem., Int. Ed. Engl. **1995**, *34*, 2039–2041. (b) Schwab, P.; Grubbs, R. H.; Ziller, J. W. J. Am. Chem. Soc. **1996**, *118*, 100–110.

⁽¹¹⁾ Byproducts derived from the catalyst were removed effectively by treatment with dimethyl sulfoxide; see: Ahn, Y. M.; Yang, K.; Georg, G. I. *Org. Lett.* **2001**, *3*, 1411–1413.

⁽¹²⁾ Increasing the concentration of 14 in benzene resulted in a lowering of the yield of 16, presumably due to the occurrence of competing oligomerization. On the other hand, the RCM reaction in dichloromethane gave decomposed materials.



^{*a*} Reagents and conditions: (a) *n*-Bu₄NF, THF, 99%; (b) Dess– Martin periodinane, CH₂Cl₂; (c) furan, *n*-BuLi, THF, 0 °C, 67% for two steps; (d) MnO₂, CH₂Cl₂, 78%; (e) L-Selectride, THF, -78°C, 95%; (f) VO(acac)₂, *t*-BuOOH, CH₂Cl₂, 0 °C, 76%; (g) TBSCl, Et₃N, 4-DMAP, CH₂Cl₂, 47% for **22** and 26% for **23**; (h) NaBH₄, CeCl₃·7H₂O, MeOH, -18 °C, 77%; (i) Ac₂O, 4-DMAP, pyridine, 89%; (j) aqueous HCl, THF, quantitative; (k) same as h, 38% and its diastereomer 28%; (l) same as i, 93%; (m) same as j, 98%; (n) MnO₂, CH₂Cl₂, 89%.

butylammonium fluoride afforded a primary alcohol derivative **19**. The Dess-Martin oxidation¹⁴ of **19** gave the aldehyde, which was allowed to react with the 2-lithiated furan to produce the furfuryl alcohols **2** and *epi-***2** as an inseparable 3:2 diastereomeric mixture regarding the carbinol center. The desired diastereomer **2** was obtained as a sole product by an oxidation-reduction procedure. Thus, the mixture of 2 and epi-2 was oxidized with manganese dioxide, and the resulting ketone 20 was reduced with L-Selectride, providing 2 stereoselectively.¹⁵ It is well recognized that the rearrangement of furfuryl alcohols to pyranones occurs under oxidation conditions, and a number of procedures for this purpose have been reported.5 We chose a vanadium-catalyzed peroxide addition followed by a rearrangement reaction,¹⁶ which was anticipated to prevent the oxidation of the diene part on the eight-membered ring. As expected, the chemoselective oxidation was realized by the treatment of 2 with tert-butyl hydroperoxide in the presence of a catalytic amount of vanadyl acetylacetonate (VO(acac)₂), eventually providing **21** as a 2:1 hemiacetal mixture. The hemiacetal hydroxyl groups in mixture 21 were protected as *tert*-butyldimethylsilyl (TBS) ethers, giving a 47% yield of α -isomer 22 and a 26% yield of β -isomer 23.¹⁷ The α -silvl ether 22 was subjected to Luche reduction conditions¹⁸ to afford allylic alcohol 24 with a high level of diastereoselectivity. Acetylation of 24 gave 25, the stereochemistry of which was confirmed by ¹H NMR analysis, including a NOE experiment.¹⁹ Acid hydrolysis of the TBS group in **25** with dilute HCl provided lactol 26. On the other hand, the minor silyl ether 23 was also transformed into 26 by the same reaction sequence used for the conversion of 22 to 26. In this case, the stereoselectivity of the Luche reduction was less satisfactory (dr = 1.3:1). Finally, the oxidation of **26** provided (\pm) mycoepoxydiene (1), the spectroscopic data (1 H and 13 C NMR) of which were well matched with those reported for natural 1.1

In conclusion, we have achieved the first total synthesis of (\pm) -mycoepoxydiene (1) in 24 steps with a 1.6% overall yield from the Diels-Alder adduct of furan and maleic anhydride. The present synthesis demonstrates the efficiency of the RCM approach to access the oxygen-bridged bicy-clooctene skeleton. Current efforts in our laboratory are directed at the development of the asymmetric synthesis of mycoepoxydiene.

Supporting Information Available: Experimental procedures and spectroscopic characterization (¹H and ¹³C NMR, IR, and HRMS) of all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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(c) Ireland, R. E.; Liu, L. J. Org. Chem. 1993, 58, 2899.

⁽¹⁵⁾ We did not confirm the configuration of the introduced stereogenic center in 2 at this stage.

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⁽¹⁷⁾ The configuration of the hemiacetal carbon in the major diastereomer **22** as depicted was determined after transformation of **22** into **25**.

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⁽¹⁹⁾ As shown in Scheme 4, significant signal enhancements at H1 (14%), H4 (7.6%), and H7 (6.9%) were observed when H5 was irradiated. The lack of signal enhancement at H6 as well as the large coupling constant ($J_{5,6} = 10.8$ Hz) indicated that H5 and H6 are in an antiperiplanar relationship. NOEs between H4/H7 (9.6%) and H5/CH₃-13 (9.3%) were also observed. Therefore, the stereochemistry of **25** was determined as depicted.